## CYCLIZATIONS OF N-(1-CHLORO-2,2,2-TRIHALOETHYLIDENE)-O-METHYL-URETHANES WITH 5-AMINO-3-METHYL-ISOXAZOLE AND 3-AMINO-5-METHYLISOXAZOLE

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*N-(1-Chloro-2,2,2-trihaloethylidene)-O-methylurethanes undergo cyclization with 5-amino-3-methylisoxazole and 3-amino-5-methylisoxazole to give respectively 6-trihalomethylisoxazolo[5,4-d]-pyrimidin-4(5H)-ones and 2-trihalomethyl-4H-isoxazolo[2,3-a]-1,3,5-triazin-4-ones.* 

**Keywords:** 5- and 3-aminoisoxazoles, N-(isoxazolyl)-N'-carbomethoxyamidines, 6-trihalomethylisoxazolo[5,4-*d*]pyrimidin-4(5H)-ones, 2-trihalomethyl-4H-isoxazolo[2,3-*a*]-1,3,5-triazin-4-ones, N-(1-chloro-2,2,2-trihaloethylidene)-O-methylurethanes, cyclization.

We have previously reported [1, 2] that the N-(1-chloro-2,2,2-trihaloethylidene)-O-methylurethanes 1a,b are convenient synthon units for the construction of trihalomethyl-substituted condensed pyrimidine systems. In particular, the reaction of the urethanes 1 with arylamines gives 2-trihalomethylquinazol-4-ones [1] and with 2-aminothiophenes the 2-trihalomethyl-3,4-dihydrothieno[2,3-*d*]pyrimidin-4-ones [2]. With the target of annelating an isoxazole ring we have continued our initial investigations and studied the reaction of the urethanes 1a,b with 5-amino-3-methylisoxazole (2a) and 3-amino-5-methylisoxazole (2b).

We have found that compounds **1a,b**, under mild conditions (benzene, 20°C) and in the presence of triethylamine, selectively imidoylate the aminoisoxazole **2a** to give the N-(isoxazol-5-yl)-N'- carbomethoxytrihaloacetamidines **3a,b**. Their <sup>1</sup>H NMR spectra show singlets for the CH<sub>3</sub>O groups (3.83 and 3.73 ppm respectively) and  $C_{(4)}$ -H protons of the isoxazole ring (6.13 and 6.03 ppm respectively). Compounds **3a,b** are of low thermal stability and when refluxed in toluene for 3 h undergo cyclization to the trihalomethyl isoxazolo[5,4-*d*]pyrimidin-4(5H)-ones **4a,b**, evidently due to an intramolecular interaction of the carbomethoxy group of the urethane fragment and the  $\pi$ -electron excessive  $C_{(4)}$  center of the isoxazole nucleus.

The IR spectra of compounds 4a,b show absorption bands for the C=O group in the range 1700-1705 cm<sup>-1</sup> and for N–H at 3170-3180 cm<sup>-1</sup>. The <sup>1</sup>H NMR spectra show the N–H protons as broad singlets in the range 11-12 ppm. Comparison of this data with the spectroscopic data for 3,4-dihydrothieno[2,3-*d*]-pyrimidin-4-ones [2] allows us to assign the compounds 4a,b a 4,5-dihydro structure. The method of synthesis proposed by us differs basically from those reported in the literature [3-6], in which bifunctional 5-amino-4-formyl(carbamoyl, thiocarbamoyl) isoxazoles are used for the construction of the pyrimidine ring.

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**1 a** Hal = F, **b** Hal = Cl; **3**, **4 a**, **c** Hal = F; **b**, **d** Hal = Cl

The reaction of the N-ethylideneurethanes 1a,b with 3-aminoisoxazole 2b occurs at room temperature similarly to the reaction with amine 2a and gives the amidines 3c,d which tend to form cyclic products when heated in refluxing *o*-xylene. However, in contrast to amidines 3a,b, the position of attack of the carbonyl group is not a carbon atom but the nitrogen atom of the isoxazole ring and this causes the formation of the

Com- pound	Empirical formula	a Found, % Calculated, % Hal N		mp, °C	Yield, %
3a*	$C_8H_8F_3N_3O_3$	$\frac{22.52}{22.69}$	$\frac{17.04}{16.73}$		71
3b	$C_8H_8Cl_3N_3O_3$	$\frac{35.59}{35.39}$	$\frac{14.17}{13.98}$	135-136 (benzene)	78
3c	$C_8H_8F_3N_3O_3$	$\frac{22.41}{22.69}$	<u>16.59</u> 16.73	69-70 (benzene–hexane, 3:1)	94
3d	$C_8H_8Cl_3N_3O_3$	$\frac{35.64}{35.39}$	$\frac{13.75}{13.98}$	168-169 (2-propanol)	92
4a	$C_{7}H_{4}F_{3}N_{3}O_{2}$	$\frac{25.80}{26.01}$	$\frac{19.05}{19.18}$	191-192 (benzene)	69
4b	$C_7H_4Cl_3N_3O_2$	$\frac{39.32}{39.61}$	$\frac{15.87}{15.65}$	168-169 (benzene)	73
4c	$C_{7}H_{4}F_{3}N_{3}O_{2}$	$\frac{26.33}{26.01}$	<u>19.43</u> 19.18	140-141 (benzene)	46
4d	$C_7H_4Cl_3N_3O_2$	<u>39.44</u> 39.61	<u>15.92</u> 15.65	215-216 (benzene)	53

TABLE 1. Characteristics of the Compounds Synthesized

\* Viscous oil, purified by reprecipitation from ether using hexane.

Com-	IR spectrum,		<sup>1</sup> H NMR spectrum, δ, ppm*	<sup>19</sup> F NMR spectrum.
pound	NH CO		1 / / 11	δ, ppm
3a	3165	1775	2.33 (3H, s, CH <sub>3</sub> ); 3.83 (3H, s, OCH <sub>3</sub> ); 6.13 (1H, s, C(4)H); 8.60 (1H, s, NH)	69.7
3b	3150	1760	2.31 (3H, s, CH <sub>3</sub> ); 3.73 (3H, s, OCH <sub>3</sub> ); 6.03 (1H, s, C(4)H); 7.68 (1H, s, NH)	
3c	3180	1765	2.46 (3H, s, CH <sub>3</sub> ); 3.84 (3H, s, OCH <sub>3</sub> ); 6.17 (1H, s, C(4)H); 9.36 (1H, s, NH)	68.7
3d	3195	1755	2.43 (3H, s, CH <sub>3</sub> ); 3.68 (3H, s, OCH <sub>3</sub> ); 6.02 (1H, s, C(4)H), 8.38 (1H, s, NH)	
<b>4</b> a	3170	1700	2.53 (3H, s, CH <sub>3</sub> ); 11.33 (1H, s, NH)	69.2
4b	3180	1705	2.52 (3H, s, CH <sub>3</sub> ); 11.65 (1H, s, NH)	
4c		1755	2.70 (3H, s, CH <sub>3</sub> ); 7.18 (1H, s, C(8)H); [2.75 (3H, s, CH <sub>3</sub> ); 6.70 (1H, s, C(8)H)]	71.1
4d		1750	2.70 (3H, s, CH <sub>3</sub> ); 7.12 (1H, s, C(8)H); [2.73 (3H, s, CH <sub>3</sub> ); 6.62 (1H, s, C(8)H)]	

TABLE 2. Spectroscopic Characteristics of Compounds 3a-d, 4a-d

\* The <sup>1</sup>H and <sup>19</sup>F NMR spectra were recorded in CDCl<sub>3</sub> (compounds **3a-d**) and DMSO-d<sub>6</sub> (compounds **4a-d**). For compounds **4c,d** the values obtained in CDCl<sub>3</sub> are given in square brackets.

2-trihalomethyl-4H-isoxazolo[2,3-*c*]-1,3,5-triazin-4-ones **4c,d**. The signals for the  $C_{(8)}$  protons appear in the <sup>1</sup>H NMR spectra in CDCl<sub>3</sub> solution at 6.70 (**4c**) and 6.62 ppm (**4d**) and in DMSO-d<sub>6</sub> solution at 7.18 (**4c**) and 7.12 ppm (**4d**) and do not disappear upon addition of water, thus confirming the presence of the triazine structure.

Compounds **4c,d** are members of the little studied condensed isoxazolo[2,3-a]-1,3,5-triazine system. There is evidence in the literature [7] for the synthesis of just isoxazolo[2,3-a]-1,3,5-triazine-2,4-diones in low yield by treating 3-aminoisoxazole with phenyl- or phenoxycarbonylisocyanates.

## EXPERIMENTAL

IR spectra were recorded on a UR-20 instrument (in  $CH_2Cl_2$  solution for compound **3a** and as KBr tablets for compounds **3b-d** and **4a-d**). <sup>1</sup>H NMR and <sup>19</sup>F spectra were taken on a Varian Gemini spectrometer (300 and 282 MHz respectively) using TMS internal standard (<sup>1</sup>H) or  $CCl_3F$  (<sup>19</sup>F).

**N-(3-Methyl-5-isoxazolyl)-N'-carbomethoxytrihaloacetamidines (3a,b)** and **N-(5-Methyl-3-isoxazolyl)-N'-carbomethoxytrihaloacetamidines (3c,d)** (Tables 1 and 2). A solution of the aminoisoxazole **2a,b** (5 mmol) and triethylamine (0.5 g, 5 mmol) in benzene (15 ml) was added with stirring at room temperature to a solution of the N-ethylideneurethane **1a,b** (5 mmol) in benzene (15 ml). After stirring for 3 h, dioxane (10 ml) was added to the reaction mixture, the precipitated triethylamine hydrochloride was filtered off, and the solvent evaporated. The product was purified by crystallization or reprecipitation.

**6-Trihalomethyl-3-methylisoxazolo**[5,4-*d*]**pyrimidin-4(5H)-ones (4a,b)** and **2-Trihalomethyl-7-methyl-4H-isoxazolo**[2,3-*a*]-1,3,5-triazin-4-ones (4c,d) (Tables 1 and 2). A solution of compound 3a,b (3 mmol) in toluene (15 ml) or of the amidine 3c,d (3 mmol) in *o*-xylene (15 ml) was heated at reflux for 3 h. Solvent was then evaporated and the residue was purified by crystallization.

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